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Enantioselective Cyclizations of Silyloxyenynes Catalyzed by Cationic Metal Phosphine Complexes

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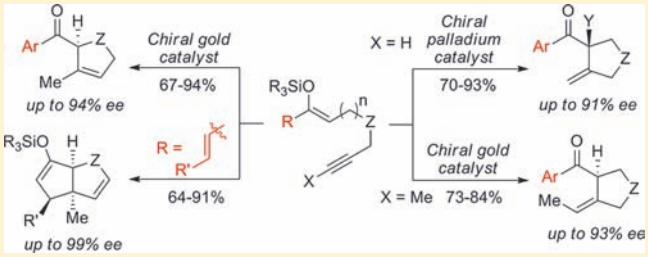
Supporting Information

ABSTRACT: The discovery of complementary methods for enantioselective transition metal-catalyzed cyclization with silyloxyenynes has been accomplished using chiral phosphine ligands. Under palladium catalysis, 1,6-silyloxyenynes bearing a terminal alkyne led to the desired five-membered ring with high enantioselectivities (up to 91% ee). As for reactions under cationic gold catalysis, 1,6- and 1,5-silyloxyenynes bearing an internal alkyne furnished the chiral cyclopentane derivatives with excellent enantiomeric excess (up to 94% ee). Modification of the substrate by incorporating an α,β-unsaturation led to the discovery of a tandem cyclization. Remarkably, using silyloxy-1,3-dien-7-ynes under gold catalysis conditions provided the bicyclic derivatives with excellent diastereo- and enantioselectivities (up to >20:1 dr and 99% ee).

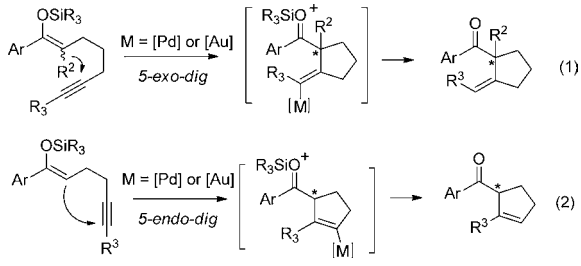


INTRODUCTION

Enantioselective α-functionalization of enolates and enolate derivatives serves as one of the most important methods for the construction of enantioenriched carbonyl containing compounds.[1] A wide range of electrophiles react with enolate derivatives, including activated carbon−carbon π-bonds.[2] In this context, alkynes are potentially interesting electrophiles, as the product of the addition reaction is an alkene that can be further elaborated. Therefore, a number of addition reactions of silyl enol ethers to alkynes have been reported.[3,4] From these reports, two major reactivity paradigms have emerged. The first, which follows from Conia's seminal report[3a] of mercury(II)-promoted addition of silyl enol ethers to alkynes, involves nucleophilic addition to the triple bond that is activated by a π-acidic transition metal complex or Lewis acid. The second more recent approach proceeds through nucleophilic addition of the enol ether to an electrophilic transition metal vinylidene generated from a terminal alkyne.[5] Despite recent developments, enantioselective variants of this class of addition reaction remain scarce. This paucity can perhaps be traced to the fact that the majority of catalysts reported for this reaction are either simple metal salts or transition metal carbonyl compounds and therefore lack a readily tunable ancillary ligand.

The past decade has witnessed the development of cationic late transition metal complexes as catalysts for addition to alkynes.[6] These complexes demonstrate the ability to catalyze the addition of nucleophiles to alkynes, even when ligated with Lewis basic phosphine ligands. For example, we reported that cationic triphenylphosphinegold(I) efficiently promoted the addition of β-ketoesters and silyl enol ethers to alkynes through a π-activation pathway.[4a]

Therefore, we envisioned that chiral phosphine analogues could catalyze such cyclization reactions in an enantioselective manner. In support of this hypothesis, Echavarren reported a gold-catalyzed enantioselective version of a 5-*exo-dig* cyclization using 1,6-enynes in the presence of methanol.[7] Moreover, related enantioselective cycloisomerization reactions have been described by Mikami[8] and Genêt[9] using catalysts based on cationic phosphinepalladium(II) and platinum(II) complexes, respectively. More recently, we,[10] Michelet,[11] and Sanz[12] showed that bisphosphinegold(I) complexes also effectively catalyzed enantioselective polycyclization and cycloisomerization reactions.[13,14] On the basis of this work, we aimed to develop a set of catalysts that would allow for enantioselective 5-*endo-dig* and 5-*exo-dig* addition reactions of enol ether nucleophiles. In this context, we focused our attention on cyclizations of 1,6- and 1,5-silyloxyenynes (eqs 1 and 2)



catalyzed by cationic (phosphine)platinum, palladium, and gold complexes as catalysts.[15] Herein, we present a full account of our studies employing chiral gold and palladium catalysts to promote asymmetric 5-*exo-dig* and 5-*endo-dig* cyclizations. This work resulted in the discoveries of complementary palladium(II)- and gold(I)-catalyzed highly enantioselective silyloxyenyne cycloisomerization reactions to yield synthetically useful exomethylencylopentane

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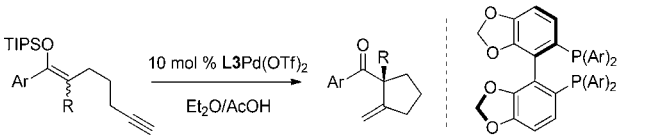
or cyclopentene derivatives. Furthermore, we also demonstrate that silyloxy-1,3-dien-7-ynes are suitable substrates for diastereo- and enantioselective cyclization reactions to form polysubstituted bicyclo[3.3.0]octane derivatives.

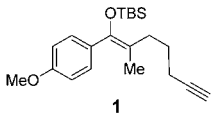
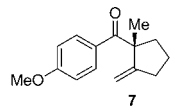
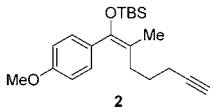
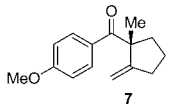
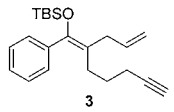
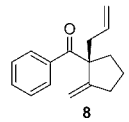
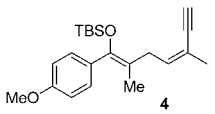
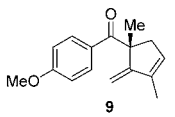
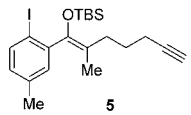
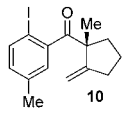
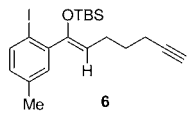
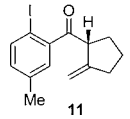
RESULTS AND DISCUSSION

Palladium-Catalyzed Enantioselective Cyclization of Silyloxy-1,6-enynes. We began our investigation with the tetrasubstituted silyl enol ether **1** depicted in Table 1. We found

Table 1. Pd-Cyclizations of Silyloxy-1,6-enynes[a]



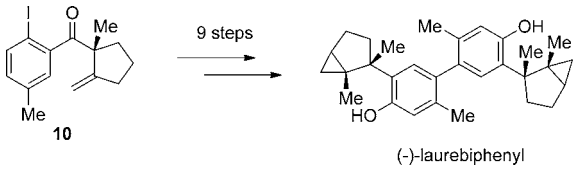
Ar = C_6H_5; (*R*)-SEGPHOS (**L1**)
Ar = 3,5-$(Me)_2$-C_6H_4; (*R*)-DM-SEGPHOS (**L2**)
Ar = 3,5-$(tBu)_2$-4-MeO-C_6H_2; (*R*)-DTBM-SEGPHOS (**L3**)

Entry	Substrate	Product	Yield (%)[b]	ee (%)[c]
1			80	78
2			93	91
3			92	88
4			70	73
5			96	95
6[d]			86	85

[a]Reactions performed at 0.02 M in Et_2O/AcOH (100/1) using 1 equiv of substrate and 10 mol % **L3**Pd$(OTf)_2$ for 16 h. [b]Isolated yields. [c]Determined by chiral HPLC. [d]At 0 °C.

that palladium complex (*R*)-DTBM-SEGPHOSPd$(OTf)_2$ [**L3**Pd$(OTf)_2$] that was successful in our previously reported enantioselective Conia-ene cyclization gave the best results in terms of enantiomeric excess (ee).[16] Indeed, (*Z*)-isomer **1** was treated with catalyst **L3**Pd$(OTf)_2$ and cyclic product **7** was isolated in 80% yield and 78% ee (entry 1). We were pleased to observe that (*E*)-isomer **2** led to the desired aryl ketone with an increase in enantioselectivity (91% ee, entry 2).

The scope of this reaction was studied, and we found that the methyl substituent could also be modified. For example, allyl-substituted ketone **8** was formed with high selectivity (88% ee, entry 3). We also observed that the 1,4-dien-6-yne **4** led to the desired cyclic ketone **9** with 73% ee (entry 4). The (*Z*)-1,6-silyloxyenyne **5** was synthesized and treated under the optimal palladium conditions to generate the desired ketone **10** with 96% yield and 95% ee (entry 5). The utility of this reaction was exemplified by the transformation of **10** into naturally occurring

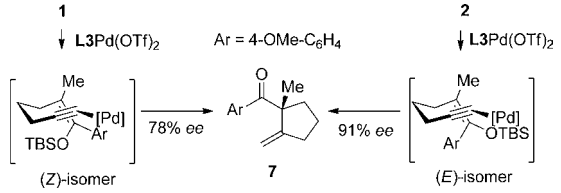


(3)

dimeric sesquiterpene (−)-laurebiphenyl (eq 3).[15] In addition, trisubstituted silyloxyenyne **6** was reacted with catalyst **L3**Pd$(OTf)_2$ at 0 °C to furnish the cyclic ketone **11**, having a tertiary stereocenter with 85% enantioselectivity (entry 6).

The high enantioselectivity obtained with **1** and **2** suggests a transition state where the chiral complex selects the face based on the placement of the large (aryl/silyloxy portion) and the small (methyl) groups of the silyl enol ether in the appropriate quadrants of the chiral environment (Scheme 1). This

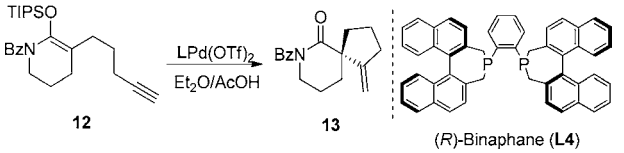
Scheme 1. Postulated Transition States



hypothesis is supported by the low impact of the alkene geometry on the enantioselectivity (Table 1, entries 1 and 2).

We decided to extend this methodology to the preparation of enantioenriched spirocyclic amides.[17] However, **L3**Pd$(OTf)_2$ with substrate **12** proved to be problematic as lower enantioselectivity was observed (Table 2, entry 1). A ligand screen was undertaken,

Table 2. Selected Optimization Experiments with Silyloxy-1,6-Enynes 12[a]



entry	ligand	conv (%)[b]	ee (%)[c]
1[d]	(*R*)-DTBM-SEGPHOS (**L3**)	15	48
2	(*R*)-DTBM-SEGPHOS (**L3**)	>95	47
3	(*R*)-SEGPHOS (**L1**)	50	88
4	(*R,S*)-SynPhos	50	84
5	(*R,S*)-CyJosiPhos	>95	5
6	(*R*)-Binaphane (**L4**)	>95 (80)[e]	98

[a]Reactions performed at 0.02 M in Et_2O/AcOH (100/1) using 1 equiv of substrate 12 and 10 mol % **L**Pd$(OTf)_2$ for 2 h. [b]Determined by ^1H NMR. [c]Determined by chiral HPLC. [d]No actetic acid was added. [e]Isolated yield.

and we found that the SEGPHOS (**L1**) ligand gave higher enantioselectivity with moderate conversion, since the hydrolysis product (corresponding lactam) was also obtained in a significant amount (entry 3). A number of bisphosphine ligands having the biaryl atropisomeric backbone were tested with moderate success.

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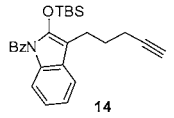
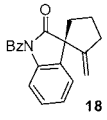
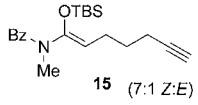
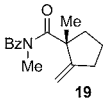
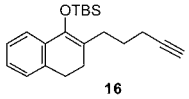
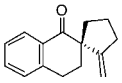
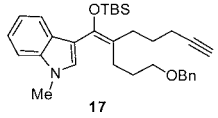
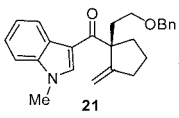
We discovered that binaphane ligand (**L4**) was a highly effective ligand with substrate **12** (entry 6). The reaction times were generally shorter and catalyst loading could be decreased substantially compared with the conditions with **L3**Pd(OTf)$_2$. Under the optimized conditions, the desired spirocyclic compound **13** was isolated with 80% yield and 98% ee.

Next, we examined the scope of this cyclization with other *O*-silylketene aminals. As shown in Table 3, 2-silyloxy indole **14**

Table 3. Pd-Cyclizations of Silyloxy-1,6-enynes[a]

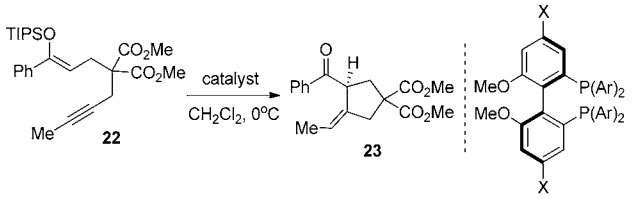
Entry	Substrate	Product	Yield (%)[b]	ee (%)[c]
1	 **14**	 **18**	83	91
2	 **15** (7:1 *Z*:*E*)	 **19**	79	80
3	 **16**	 **20**	91	87
4	 **17**	 **21**	80	89

[a]Reactions performed at 0.02 M in CH_2Cl_2/AcOH (100/1) using 1 equiv of substrate and 5 mol % **L4**Pd(OTf)$_2$ for 120 min. [b]Isolated yields. [c]Determined by chiral HPLC.

afforded the desired spiro-oxindole product **18** with 83% yield and 91% ee (entry 1). The acyclic *O*-silylketene aminal **15** was also reacted under similar conditions to obtain amide **19** with satisfactory enantioselectivity (entry 2). The efficiency of this catalyst is particularly noteworthy, as treatment of silyl enol ethers **16** and **17** with **L4**Pd(OTf)$_2$ gave the corresponding cyclopentane adducts with high enantioselectivity (entries 3 and 4), while attempts at the **L3**Pd(OTf)$_2$-catalyzed cyclization of these substrates provided trace amounts of the desired products.[18]

Gold-Catalyzed Enantioselective Cyclization of Silyloxy-1,6-enynes. During the course of our study of this enantioselective palladium-catalyzed 5-*exo*-*dig* cyclization reaction, we found that substrates bearing an internal alkyne such as **22** were not reactive with catalyst **L3**Pd(OTf)$_2$ (Table 4, entry 1). With the more reactive binaphane ligand (**L4**), the ketone derived from the hydrolysis of the silyl enol ether was obtained as the major product (entry 2). We also tested substrate **22** under platinum catalysis with **L3**Pt(OTf)$_2$, and no reaction was observed (entry 3). Despite little success with chiral cationic gold complexes and substrates bearing a terminal alkyne,[19] we decided to explore the reactivity of **22** with a different set of ligands under gold catalysis. An initial screen revealed that **L3**(AuCl)$_2$ could promote the desired 5-*exo*-*dig* cylization to give **23**, albeit with low yield and enantioselectivity (entries 4 and 5).[20] We subjected **22** to previously reported conditions for the 6-*exo*-*dig* cyclization[10b] (**L5**(AuCl)$_2$ and $AgSbF_6$), and moderate yields (42%) of desired cyclized product **23** were obtained with a slight improvement in the enantioselectivity (entry 6). The low conversions were associated with the hydrolysis of the enolsilane, probably due

Table 4. Selected Optimization Experiments with Silyloxy-1,6-enynes[a]



Ar = 3,5-(tBu)$_2$-C$_6$H$_3$; X = H; (*R*)-MeO-DTB-BIPHEP (**L5**)
Ar = 3,5-(tBu)$_2$-4-MeO-C$_6$H$_2$, X = H; (*R*)-MeO-DTBM-BIPHEP (**L6**)
Ar = 3,5-(tBu)$_2$-4-MeO-C$_6$H$_2$, X = OMe; (*R*)-DTBM-GARPHOS (**L7**)

entry	catalyst	additive	yield (%)[b]	ee (%)[c]
1	**L3**Pd(OTf)$_2$	none	NR	ND
2	**L4**Pd(OTf)$_2$	none	traces	ND
3	**L3**Pt(OTf)$_2$	none	NR	ND
4	**L3**(AuCl)$_2$	AgOTf	39[d]	17
5	**L3**(AuCl)$_2$	$AgSbF_6$	37[d]	32
6	**L5**(AuCl)$_2$	$AgSbF_6$	42[d]	35
7	**L5**(AuCl)$_2$	NaBARF	67	52
8	**L6**(AuCl)$_2$	NaBARF	81	72
9[e]	**L6**(AuCl)$_2$	NaBARF	74	85
10[e,f]	**L6**(AuCl)$_2$	NaBARF	84	93
11[e,f]	**L7**(AuCl)$_2$	NaBARF	88	87

[a]Reactions performed at 0.1 M using 1 equiv of **22**, 5 mol % catalyst, and 10 mol % additive for 16 h. NR and ND mean no reaction and not determined, respectively. [b]Isolated yields. [c]Determined by chiral HPLC. [d]Determined ^{1}H NMR versus using an internal standard (diethyl phthalate). [e]Reaction was performed at −30 °C. [f]Using 1,2-dichloroethane (DCE) as a solvent.

to trace amounts of acid formed under the reaction conditions with silver salts.[21] Performing the reaction at low temperature and using sodium tetrakis[3,5-bis(trifluoromethyl)phenyl]borate (NaBARF) as a chloride scavenger in combination with **L5**(AuCl)$_2$ led to the desired cyclized ketone **23** in 67% yield and 52% ee (entry 7). Lowering the temperature and switching to **L6**(AuCl)$_2$ was promising in terms of enantioselectivity (entries 8 and 9). Upon further optimization of the reaction parameters, the desired product was isolated in 84% yield and 93% ee when the reaction was performed at −30 °C using dichloroethane as the solvent (entry 10).

We explored the optimized conditions with substrates having different substituents on the aryl moiety. As shown in Table 5, the cyclization proceeded cleanly with substrates bearing methyl substitution (**24**, **25**, and **26**) to give the desired products with high enantiomeric excess (86−91% ee). Having a chlorine substituent such as in **28** was deleterious to reactivity, and a higher reaction temperature was required (entry 5). Finally, modifying the position of the *gem*-diester as in substrate **29** had a significant impact on the enantioselectivity, and the cyclic product **35** was obtained with 50% ee (entry 6).

Gold-Catalyzed Enantioselective Cyclization of Silyloxy-1,5-enynes. Decreasing the tether length by one carbon allowed us to examine the 5-*endo*-*dig* process in further detail. We suspected that silyloxy-1,5-enynes such as **36** would be suitable substrates and could favor only the product derived from the 5-*endo* attack. We initially subjected **36** to palladium and platinum catalysis; however reaction catalyzed by these d^8 metal complexes gave none of the desired product (Table 6, entries 1 and 2). Treatment of the same substrate with gold catalyst **L6**(AuCl)$_2$ and NaBARF gave the desired product **37** but with low enantioselectivity (entry 3). The catalyst derived from (*R*)-SEGPHOS(AuCl)$_2$ [**L1**(AuCl)$_2$] and NaBARF was

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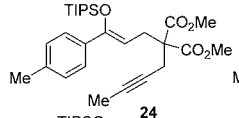
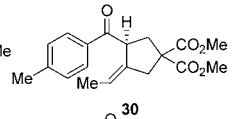
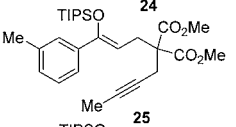
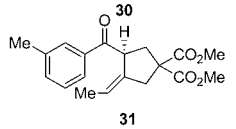
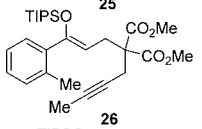
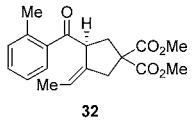
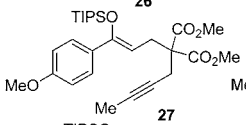
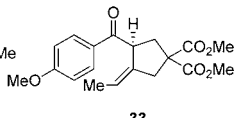
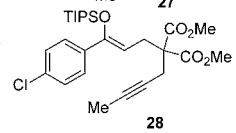
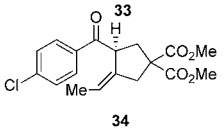
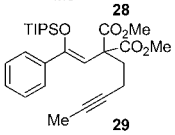
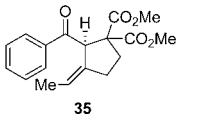
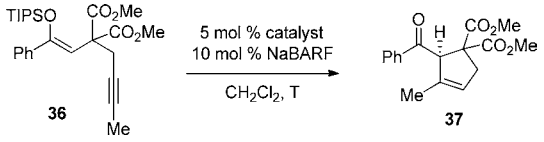
Table 5. Enantioselective Au(I)-Cyclizations with Silyloxy-1,6-Enynes[a]

Entry	Substrate	Product	Yield (%)[b]	ee (%)[c]
1	 **24**	 **30**	83	86
2	 **25**	 **31**	86	90
3	 **26**	 **32**	88	91
4	 **27**	 **33**	81	79
5[d]	 **28**	 **34**	73	71
6[e]	 **29**	 **35**	79[f]	50

[a]Reactions performed at 0.1 M using 1 equiv of substrate, 5 mol % (*R*)-MeO-DTBM-BIPHEP(AuCl)$_2$ [**L5**(AuCl)$_2$], and 10 mol % NaBARF for 16 h in DCE at −30 °C. [b]Isolated yields. [c]Determined by chiral HPLC. [d]Reaction was performed at −10 °C. [e]Using dichloromethane as a solvent. [f]Isolated as an inseparable, 4:1 mixture of cyclic products.

Table 6. Selected Optimization Experiments with Silyloxy-1,5-enynes[a]

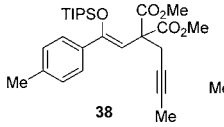
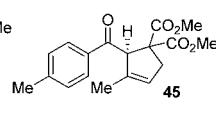
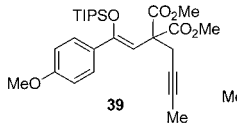
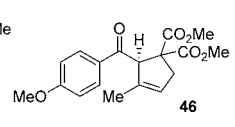
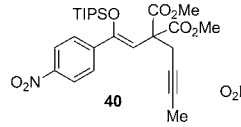
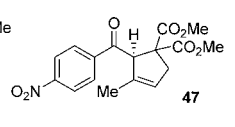
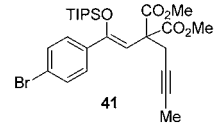
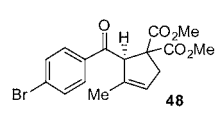
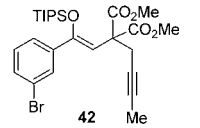
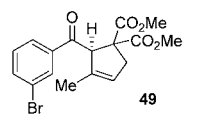
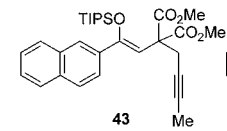
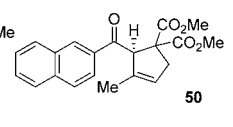
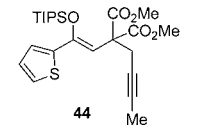
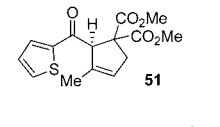


entry	catalyst	*T* (°C)	yield (%)[b]	ee (%)[c]
1[d]	**L3**Pd(OTf)$_2$	r.t.	NR	ND
2[d]	**L3**Pt(OTf)$_2$	r.t	NR	ND
3	**L6**(AuCl)$_2$	r.t.	76	24
4	**L1**(AuCl)$_2$	r.t.	69	4
5	**L2**(AuCl)$_2$	r.t.	72	28
6	**L3**(AuCl)$_2$	r.t.	74	73
7	**L3**(AuCl)$_2$	−10	79	82
8	**L3**(AuCl)$_2$	−30	81	88
9	**L3**(AuCl)$_2$	−50	75	94

[a]Reactions performed at 0.1 M using 1 equiv of **36**, 5 mol % catalyst, and 10 mol % additive for 16 h. [b]Isolated yields. [c]Determined by chiral HPLC. [d]No NaBARF was added during this reaction.

also tested and furnished the desired product with no significant enantiomeric excess (entry 4). However, increasing the size of the substituents on the phosphorus aryl moieties proved to be beneficial for enantioselectivity and led to an encouraging 73% ee (entry 6). Moreover, performing the cyclization using **L3**(AuCl)$_2$ and NaBARF at lower temperature (−50 °C) greatly improved the selectivity, as **37** was isolated with 94% ee and 75% yield (entry 9).

Table 7. Enantioselective Au(I)-Cyclization with Silyloxy-1,5-Enynes[a]

Entry	Substrate	Product	Yield (%)[b]	ee (%)[c]
1	 **38**	 **45**	81	91
2	 **39**	 **46**	92	94
3[e]	 **40**	 **47**	71	94
4	 **41**	 **48**	67	92
5	 **42**	 **49**	72	93
6	 **43**	 **50**	82	89
7	 **44**	 **51**	94	90

[a]Reactions performed at 0.1 M using 1 equiv of substrate, 5 mol % (*R*)-DTBM-SEGPHOS(AuCl)$_2$ [**L3**(AuCl)$_2$], and 10 mol % NaBARF for 16 h at −30 °C. [b]Isolated yields. [c]Determined by chiral HPLC. [e]Using 1,2-dichloroethane as a solvent.

With these results in hand, we next sought to evaluate the substrate scope of our optimized conditions (Table 7). Various substituted phenyl derived silyl enol ethers (**38**−**42**) furnished the cyclized products (**45**−**49**) in good yields (67−92%) and high enantiomeric excess (91−94% ee) (entries 1−5). The reactions with substrates bearing an electron-poor aryl substituent showed lower reaction rates but still provided the desired product in high enantiomeric purity. Similarly, 2-naphthyl (**43**) and 2-thiophenyl (**44**) substituted substrates reacted under these conditions to generate the desired ketones in high yields and enantioselectivities (entries 6 and 7, 89 and 90% ee, respectively). The absolute stereochemistry of the cyclopentene derivatives was assigned by analogy to an X-ray structure obtained after recrystallization of aryl ketone **47**.[22]

As depicted in Table 8, decreasing the size of the substituents on the silyl group (**52**, TES and **53**, TBDMS) was slightly detrimental in terms of yields but still led to good enantioselectivities (87% and 86% ee, respectively). Next, we observed that substrate **54**, having no substitution on the backbone, gave lower enantioselectivity (entry 4, 55% ee). Treatment of malononitrile derivative **55** under the optimal conditions gave **58** with an increase in enantioselectivity (entry 5, 71% ee).

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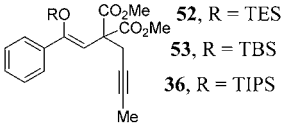
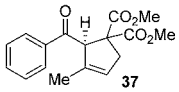
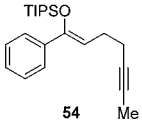
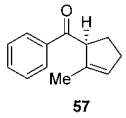
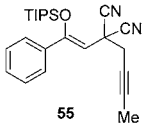
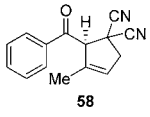
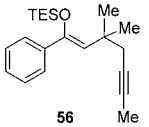
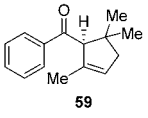
Table 8. Enantioselective Au(I)-Cyclization with Silyloxy-1,5-Enynes[a]

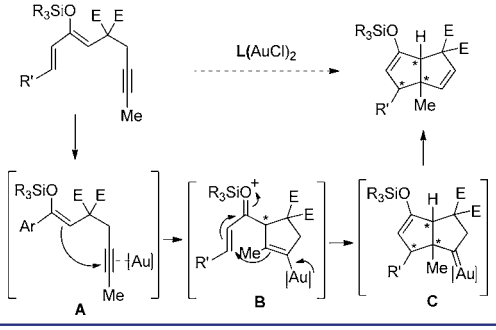
Entry	Substrate	Product	Yield (%)[b]	ee (%)[c]
1	 **52**, R = TES	 **37**	71	86
2	**53**, R = TBS		67	87
3	**36**, R = TIPS		75	94
4	 **54**	 **57**	77	55
5[e]	 **55**	 **58**	58	71
6	 **56**	 **59**	81	90

[a]Reactions performed at 0.1 M using 1 equiv of substrate, 5 mol % (*R*)-DTBM-SEGPHOS(AuCl)$_2$ [**L3**(AuCl)$_2$], and 10 mol % NaBARF for 16 h at −30 °C. [b]Isolated yields. [c]Determined by chiral HPLC. [e]Using 1,2-dichloroethane as a solvent.

Finally, we were pleased to find that the gold-catalyzed cyclization of **56** led to the desired dimethyl-substituted product **59** in 81% yield and 90% ee (entry 6).

Gold-Catalyzed Enantioselective Tandem Cyclization of Silyloxy-1,3-dien-7-yne. Subsequently, as depicted in Scheme 2, we explored the reactivity of the analogous 3-siloxy-1,

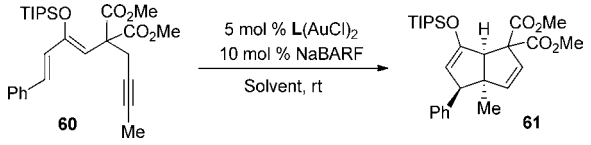
Scheme 2. Proposed Intermediates for Gold-Catalyzed Tandem Cyclization of Silyloxy-1,3-dien-7-yne



3-diene-7-yne toward gold(I) catalysis, anticipating the formation of bicyclo[3.3.0]octane derivatives. For this scenario to be successful, the vinyl gold intermediate **B** obtained after the cyclization would perform a conjugate addition on the activated unsaturated ketone to form an additional carbon−carbon bond.[23] Under this proposed mechanism, the carbene intermediate[24] would undergo subsequent 1,2-hydrogen migration to give the desired bicylic diene.[25]

To our delight, the reaction performed with substrate **60** and **L3**(AuCl)$_2$ gave the desired product **61** with excellent diastereoselectivity and enantioselectivity (Table 9, entry 1). In this case, the reaction was performed at room temperature and dry molecular sieves were added to the reaction mixture in order to avoid the formation of a complex mixture of ketones. The optimized conditions using dichloroethane as a solvent

Table 9. Selected Optimization for Enantioselective Au(I)-Cyclization with Silyloxy-1,3-dien-7-ynes[a]



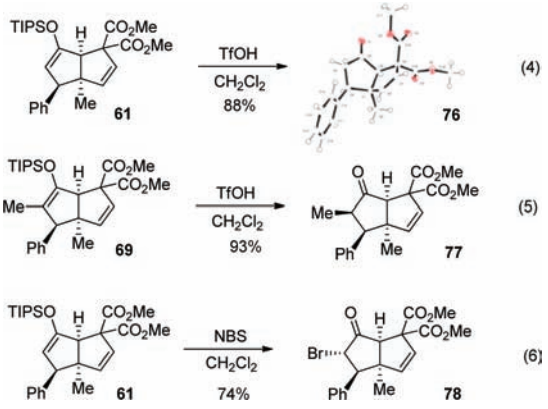
entry	ligand	solvent	dr[b]	yield (%)[c]	ee (%)[d]
1	**L3**	CH_2Cl_2	>20:1	76	98
2	**L3**	DCE	>20:1	91	99
3	**L1**	CH_2Cl_2	>20:1	36	41
4	**L2**	CH_2Cl_2	>20:1	72	56

[a]Reactions performed at 0.1 M using 1 equiv of **60**, 5 mol % **L3**(AuCl)$_2$, and 10 mol % NaBARF for 16 h. [b]The diastereoselectivity was determined by ^{1}H NMR of the crude reaction mixture. [c]Isolated yields. [d]Determined by chiral HPLC.

(entry 2) gave **61** in a very impressive 99% ee, with excellent yield (91%) and diastereoselectivity (>20:1). We noted that both the electron density and steric hindrance associated with bulky ligand **L3** were essential to obtain excellent yields and enantioselectivities (entries 3 and 4).

This tandem gold(I)-catalyzed process showed great compatibility with the presence of a wide range of substrates (Table 10). Tetrasubstituted diene **62** gave the desired bicyclic product **69** with high yield (76%) and enantioselectivity (95% ee, entry 1). Modification of the terminal substituent of the alkyne to an ethyl group gave the desired silyl enol ether **63** in a satisfactory yield (61%) and enantioselectivity (96% ee, entry 2). The reactions were faster with substrates **64** and **65** and resulted in the formation of the bicyclic dienes**71** and **72** with high enantioselectivities (98% and 89% ee, entries 3 and 4). However, substrate having a *para*-methoxy substitution on the aryl ring gave a complex mixture of products. We found that the reaction was efficient with a substrate bearing the 1-naphthyl group (**66**); the bicyclic product **73** was obtained with high enantioselectivity (91% ee, entry 5). Finally, the reaction was found to tolerate various alkyl substituents at R^1 (*c*-hexyl and *n*-butyl) and gave the bicyclic ketones **74** and **75** in good yield with slightly lower enantioselectivites (73 and 81% ee, entries 6 and 7).

To exemplify the utility of this tandem cyclization, silyl enol ether **61** was hydrolyzed in the presence of acid to give the corresponding ketone **76** (eq 4). The absolute stereochemistry was determined by X-ray analysis of this crystalline compound,



and the stereochemistry of the related bicyclo[3.3.0]octane derivatives was assigned by analogy. Ketone **77**, containing four contiguous stereogenic centers, was obtained as a single stereoisomer by treatment of **69** under similar conditions (eq 5).

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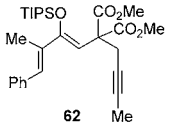
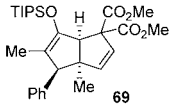
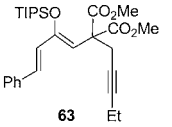
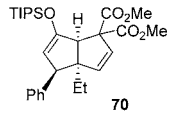
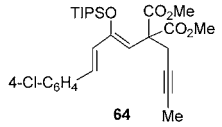
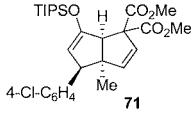
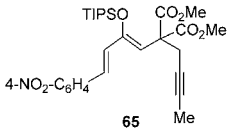
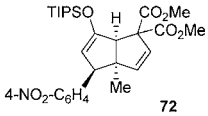
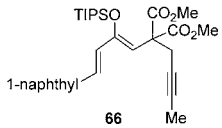
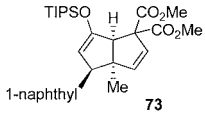
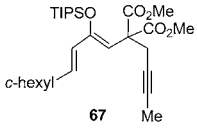
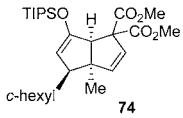
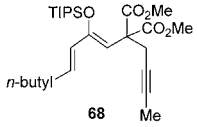
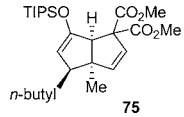
Table 10. Enantioselective Au(I)-Cyclization with Silyloxy-1,3-dien-7-ynes[a,b]

Entry	Substrate	Product	Yield (%)[c]	ee (%)[d]
1	 **62**	 **69**	76	95
2	 **63**	 **70**	61	96
3	 **64**	 **71**	64	98
4	 **65**	 **72**	81	89
5	 **66**	 **73**	72	91
6	 **67**	 **74**	68	81
7	 **68**	 **75**	82	73

[a]Reactions performed at 0.1 M using 1 equiv of substrate, 5 mol % (*R*)-DTBM-SEGPHOS(AuCl)$_2$, and 10 mol % NaBARF for 16 h. [b]The diastereoselectivity was >20:1 as determined by ^{1}H NMR of the crude reaction mixture. [c]Isolated yields. [d]Determined by chiral HPLC.

Furthermore, bromination of **61** afforded the α-bromoketone **78** with high diastereoselectivity (>20:1, eq 6). The preference for formation of the *anti*-substituted product may be explained by increased steric repulsion between the β-substituent and the methyl group in transition state **80**, leading to the *syn*-substituted adduct (Figure 1).

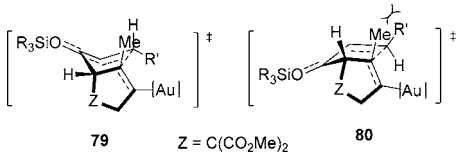


Figure 1. Rationale for *trans* Diastereoselectivity.

CONCLUSION

In summary, we have described novel asymmetric metal-catalyzed 5-*exo* and 5-*endo-dig* cyclizations of syliloxyenynes using palladium and gold complexes as catalysts. These reactions showed excellent enantioselectivity and provided entry into a wide range of cyclopentanoid structures. While the palladium- and gold-catalyzed cyclization reactions are mechanistically related, the two catalyst have complementary limitations and advantages; the palladium(II) complexes were generally limited to catalysis of 5-*exo*-dig cyclization reactions of terminal alkynes, while the gold(I) catalysts showed preference for cyclization reactions of nonterminal alkynes. Moreover, the chiral phosphine gold complexes provided access to enantioselective 5-*endo*-dig cyclization reactions previously unachievable through catalysis with cationic group 10 metal complexes. Taken together, these results further highlight the great potential of electrophilic late transition metal complexes to serve as catalysts for enantioselective formation of carbon–carbon bonds by alkyne activation.

ASSOCIATED CONTENT

Supporting Information

Experimental procedures and compound characterization cif data. This material is available free of charge via the Internet at http://pubs.acs.org.

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REFERENCES

(1) Selected examples of catalytic enantioselective synthesis of ketones: (a) Evans, P. A.; Leahy, D. K. *J. Am. Chem. Soc.* **2003**, *125*, 8974. (b) Behenna, D. C.; Stoltz, B. M. *J. Am. Chem. Soc.* **2004**, *126*, 15044. (c) Doyle, A. G.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2005**, *127*, 62. (d) Trost, B. M.; Xu, J. *J. Am. Chem. Soc.* **2005**, *127*, 2846. (e) Yan, X.-X.; Liang, C.-H.; Zhang, Y.; Hong, W.; Cao, B.-X.; Dai, L.-X.; Hou, X.-L. *Angew. Chem., Int. Ed.* **2005**, *44*, 6544. (f) Weix, D. J.; Hartwig, J. F. *J. Am. Chem. Soc.* **2007**, *129*, 7720. (g) Lundin, P. M.; Fu, G. C. *J. Am. Chem. Soc.* **2010**, *132*, 11027. (h) Mastracchio, A.; Warkentin, A. A.; Walji, A. M.; MacMillan, D. W. C. *Proc. Natl. Acad. Sci. U.S.A.* **2010**, *107*, 20648. (i) Cheon, C. H.; Kanno, O.; Toste, F. D. *J. Am. Chem. Soc.* **2011**, *133*, 13248. For examples of enantioselective α-vinylation of carbonyl compounds, see: (j) Kim, K.; MacMillan, D. W. C. *J. Am. Chem. Soc.* **2008**, *130*, 398. (k) Taylor, A. M.; Altman, R. A.; Buchwald, S. L. *J. Am. Chem. Soc.* **2009**, *131*, 9900.

(2) For a review of metalenolate additions to carbon–carbon multiple bonds, see: Denes, F.; Perez-Luna, A.; Chemla, F. *Chem. Rev.* **2010**, *110*, 2366.

(3) For metal-catalyzed cyclization reactions of alkynyl silyl enol ethers, see the following: For *mercury*: (a) Drouin, J.; Boaventura, M.-A.; Conia, J.-M. *J. Am. Chem. Soc.* **1985**, *107*, 1726. (b) Drouin, J.; Boaventura, M.-A. *Tetrahedron Lett.* **1987**, *28*, 3923. (c) Huang, H.; Forsyth, C. J. *J. Org. Chem.* **1995**, *60*, 2773. (d) Imamura, K.; Yoshikawa, E.; Gevorgyan, V.; Yamamoto, Y. *Tetrahedron Lett.* **1999**, *40*, 4081. For *tungsten*: (e) Maeyama, K.; Iwasawa, N. *J. Am. Chem.*

Soc. **1998**, *120*, 1928. (f) Iwasawa, N.; Maeyama, K.; Kusama, H. *Org. Lett.* **2001**, *3*, 3871. (g) Kusama, H.; Yamabe, H.; Iwasawa, N. *Org. Lett.* **2002**, *4*, 2569. (h) Iwasawa, N.; Miura, T.; Kiyota, K.; Kusama, H.; Lee, K.; Lee, P. H. *Org. Lett.* **2002**, *4*, 4463. (i) Miura, T.; Kiyota, K.; Kusama, H.; Lee, K.; Kim, H.; Kim, S.; Lee, P. H.; Iwasawa, N. *Org. Lett.* **2003**, *5*, 1725. (j) Kusama, H.; Onizawa, Y.; Iwasawa, N. *J. Am. Chem. Soc.* **2006**, *128*, 16500. (k) Grandmarre, A.; Kusama, H.; Iwasawa, N. *Chem. Lett.* **2007**, *36*, 66. For *rhodium* (l) Dankwardt, J. W. *Tetrahedron Lett.* **2001**, *42*, 5809. For *rhenium* (m) Kusawa, H.; Yamabe, H.; Onizawa, Y.; Hoshino, T.; Iwasawa, N. *Angew. Chem., Int. Ed.* **2005**, *44*, 468. (n) Saito, K.; Onizawa, Y.; Kusama, H.; Iwasawa, N. *Chem.—Eur. J.* **2010**, *16*, 4716. For *platinum* (o) Nevado, C.; Cardenas, D. J.; Echavarren, A. M. *Chem.—Eur. J.* **2003**, *9*, 2627. For *silver* (p) Godet, T.; Belmont, P. *Synlett* **2008**, 2513.

(4) For gold-catalyzed cyclization reactions of alkynyl silyl enol ethers, see: (a) Staben, S. T.; Kennedy-Smith, J. J.; Huang, D.; Corkey, B. K.; LaLonde, R. L.; Toste, F. D. *Angew. Chem., Int. Ed.* **2006**, *45*, 5991. (b) Lee, K.; Lee, P. H. *Adv. Synth. Catal.* **2007**, *349*, 2092. (c) Minnihan, E. C.; Colletti, S. L.; Toste, F. D.; Shen, H. C. *J. Org. Chem.* **2007**, *72*, 6287. (d) Barabe, F.; Betournay, G.; Bellavance, G.; Barriault, L. *Org. Lett.* **2009**, *11*, 4236. (e) Kusama, H.; Karibe, Y.; Onizawa, Y.; Iwasawa, N. *Angew. Chem., Int. Ed.* **2010**, *49*, 4269. (f) Ito, H.; Ohmiya, H.; Sawamura, M. *Org. Lett.* **2010**, *12*, 4380. (g) Hashmi, A. S. K.; Yang, W.; Rominger, F. *Angew. Chem., Int. Ed.* **2011**, *50*, 5762. (h) Barabe, F.; Levesque, P.; Korobkov, I.; Barriault, L. *Org. Lett.* **2011**, *13*, 5580. See also: (i) Shapiro, N. D.; Toste, F. D. *Proc. Natl. Acad. Sci.* **2008**, *105*, 2779.

(5) (a) Maeyama, K.; Iwasawa, N. *J. Am. Chem. Soc.* **1998**, *120*, 1928. (b) Iwasawa, N.; Maeyama, K.; Kusama, H. *Org. Lett.* **2001**, *3*, 3871. (c) Kusama, H.; Yamabe, H.; Iwasawa, N. *Org. Lett.* **2002**, *4*, 2569. (d) Iwasawa, N.; Miura, T.; Kiyota, K.; Kusama, H.; Lee, K.; Lee, P. H. *Org. Lett.* **2002**, *4*, 4463.

(6) For recent reviews on cycloisomerization reactions: (a) Buisine, O.; Aubert, C.; Malacria, M. *Chem. Rev.* **2002**, *102*, 813. (b) Lloyd-Jones, G. *Org. Biomol. Chem.* **2003**, *1*, 215. (c) Fairlamb, I. J. S. *Angew. Chem., Int. Ed.* **2004**, *43*, 1048. (d) Zhang, L.; Sun, J.; Kozmin, S. *Adv. Synth. Catal.* **2006**, *348*, 2271. (e) Ma, S.; Yu, S.; Gu, Z. *Angew. Chem., Int. Ed.* **2006**, *45*, 200. (f) Michelet, V.; Toullec, P. Y.; Genêt, J.-P. *Angew. Chem., Int. Ed.* **2008**, *47*, 4727. (g) Jiménez-Nùñez, E.; Echavarren, A. M. *Chem. Rev.* **2008**, *108*, 3326. (h) Furstner, A. *Chem. Soc. Rev.* **2009**, *38*, 3208. (i) Belmont, P.; Parker, E. *Eur. J. Org. Chem.* **2009**, *35*, 6075. (j) Lee, S. I.; Chatani, N. *Chem. Commun.* **2009**, 371. (k) Trost, B. M.; Gutierrez, A. C.; Ferreira, E. F. *J. Am. Chem. Soc.* **2010**, *132*, 9206. (l) Toullec, P. Y.; Michelet, V. *Top. Curr. Chem.* **2011**, *302*, 31.

(7) Munoz, M. P.; Adrio, J.; Carrtero, J. C.; Echavarren, A. M. *Organometallics* **2005**, *24*, 1293.

(8) Hatano, M.; Terada, M.; Mikami, K. *Angew. Chem., Int. Ed.* **2001**, *40*, 249.

(9) (a) Charruault, L.; Michelet, V.; Taras, R.; Gladiali, S.; Genêt, J.-P. *Chem. Commun.* **2004**, 850. (b) Toullec, P. Y.; Chao, C.-M.; Chen, Q.; Gladiali, S.; Genet, J. P.; Michelet, V. *Adv. Synth. Catal.* **2008**, *350*, 2401.

(10) Enantioselective gold-catalyzed 6-*exo*-*dig* cyclizations: (a) Sethofer, S. G.; Staben, S. T.; Hung, O. Y.; Toste, F. D. *Org. Lett.* **2008**, *10*, 4315. (b) Sethofer, S. G.; Mayer, T.; Toste, F. D. *J. Am. Chem. Soc.* **2010**, *132*, 8276.

(11) (a) Chao, C.-M.; Viatle, M. R.; Toullec, P. Y.; Genêt, J.-P.; Michelet, V. *Chem.—Eur. J.* **2009**, *15*, 1319. (b) Chao, C.-M.; Genin, E.; Toullec, P. Y.; Genêt, J.-P.; Michelet, V. *J. Organomet. Chem.* **2009**, *694*, 538. (c) Chao, C.-M.; Beltrami, D.; Toullec, P. Y.; Michelet, V. *Chem. Commun.* **2009**, 6988. (d) Pradal, A.; Chao, C.-M.; Vitale, M. R.; Toullec, P. Y.; Michelet, V. *Tetrahedron* **2011**, *67*, 4371.

(12) Martínez, A.; García-García, P.; Fernández-Rodríguez, M. A.; Rodríguez, F.; Sanz, R. *Angew. Chem., Int. Ed.* **2010**, *49*, 4633.

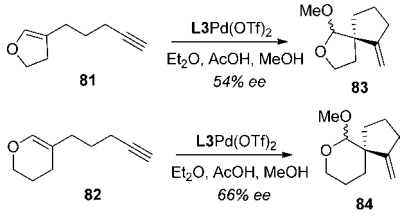
(13) For recent reviews on gold catalysis: (a) Gorin, D. J.; Sherry, B. D.; Toste, F. D. *Chem. Rev.* **2008**, *108*, 3351. (b) Jimenez-Nunez, E.; Echavarren, A. M. *Chem. Rev.* **2008**, *108*, 3326. (c) Arcadi, A. *Chem. Rev.* **2008**, *108*, 3266. (d) Das, A.; Abu, S. M. A.; Liu, R.-S. *Org. Biomol. Chem.* **2010**, *8*, 960.

(14) For recent reviews on enantioselective gold-catalyzed transformations: (a) Bongers, N.; Krause, N. *Angew. Chem., Int. Ed.* **2008**, *47*, 2178. (b) Widenhoefer, R. A. *Chem.—Eur. J.* **2008**, *14*, 5382. (c) Shapiro, N.; Toste, F. D. *Synlett* **2010**, *5*, 675. (d) Sengupta, S.; Shi, X. *ChemCatChem* **2010**, *2*, 609. (e) Lee, A.-L. *Annu. Rep. Prog. Chem., Sect. B: Org. Chem.* **2010**, *106*, 428. (f) Pradal, A.; Toullec, P. Y.; Michelet, V. *Synthesis* **2011**, *10*, 1501.

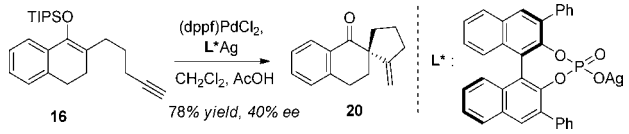
(15) For a preliminary account describing Pd-catalyzed enantioselective cyclization with alkynyl silyl enol ether, see: Corkey, B. K.; Toste, F. D. *J. Am. Chem. Soc.* **2007**, *129*, 2764.

(16) Corkey, B. K.; Toste, F. D. *J. Am. Chem. Soc.* **2005**, *127*, 17168.

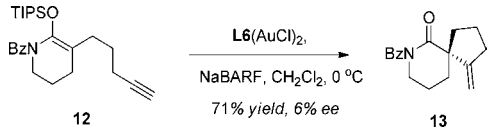
(17) We also studied the reactivity of cyclic enol ethers **81** and **82** to generate *spiro*-cyclic ethers. However, moderate yields and enantioselectivities were noticed under our optimized conditions.



(18) Cyclization reaction of silyl enol ether **16** with achiral (phosphine)palladium complexes in the presence of chiral counterions gave **20** with up to 40% ee. For an additional example of this strategy, see: (a) Hamilton, G. L.; Kang, E. J.; Mba, M.; Toste, F. D. *Science* **2007**, *317*, 496. (b) Mukherjee, A.; List, B. *J. Am. Chem. Soc.* **2007**, *129*, 11336. (c) LaLonde, R. L.; Wang, Z. J.; Mba, M.; Lackner, A. D.; Toste, F. D. *Angew. Chem., Int. Ed.* **2010**, *49*, 598. (d) Jaing, G.; Halder, R.; Fang, Y.; List, B. *Angew. Chem., Int. Ed.* **2011**, *50*, 9752. (e) Rauniyar, V.; Wang, Z. J.; Burks, H. E.; Toste, F. D. *J. Am. Chem. Soc.* **2011**, *133*, 8486. (f) Barbazanges, M.; Augé, M.; Moussa, J.; Amouri, H.; Aubert, C.; Desmarets, C.; Fensterbank, L.; Gandon, V.; Malacria, M.; Olliver, C. *Chem.—Eur. J.* **2011**, *48*, 13789.



(19) Various silyl enol ethers having a terminal alkyne were tested under gold catalysis with little success in terms of enantioselectivity. The experiment below with substrate **12** highlights the complementary utility of palladium (Table 2, entry 6) and gold catalysis.



(20) The cyclization of silyloxyenyne **22** was also tested with chiral phosphoramidite and (acyclic diamino)carbene gold(I) complexes in the presence of NaBARF, and low enantioselectivities were obtained (21% and 6% ee, respectively). For details of these catalysts: *phosphoramidites*: (a) Alonso, I.; Trllo, B.; López, F.; Montserrat, S.; Ujaque, G.; Caseto, L.; Lledos, A.; Masareñas, J. L. *J. Am. Chem. Soc.* **2009**, *131*, 13020. (b) Gonzalez, A. Z.; Toste, F. D. *Org. Lett.* **2010**, *12*, 200. (c) Teller, H.; Flugge, S.; Goddard, R.; Furstner, A. *Angew. Chem., Int. Ed.* **2010**, *49*, 1949. (d) Gonzalez, A. Z.; Benitez, D.; Tkatchouk, E.; Goddard, W. A.; Toste, F. D. *J. Am. Chem. Soc.* **2011**, *133*, 5500. *(acyclic diamino)carbene:* (e) Wang, Y.-M.; Kuzniewski, C. N.; Rauniyar, V.; Hoong, C.; Toste, F. D. *J. Am. Chem. Soc.* **2011**, *133*, 12972.

(21) Treatment of **22** with AgOTf or TfOH led to the corresponding hydrolyzed ketone in small amounts.

(22) See Supporting Information.

(23) For references on all-carbon quaternary centers, see: (a) *Quaternary Stereocenters: Challenges and Solutions for Organic Synthesis*; Christophers, J., Baro, A., Eds.; Wiley-VCH: Weinheim, 2006. (b) Douglas, C. J.; Overman, L. E. *Proc. Natl. Acad. Sci. U.S.A.* **2004**, *101*, 5363. (c) Cozzi, P. G.; Hilgraf, R.; Zimmermann, N. *Eur. J. Org. Chem.* **2007**, 5969.

(24) (a) Seidel, G.; Mynott, R.; Furstner, A. F. *Angew. Chem., Int. Ed.* **2009**, *48*, 2510. (b) Benitez, D.; Shapiro, N. D.; Tkatchouk, E.; Wang, Y.; Goddard, W. A.; Toste, F. D. *Nat. Chem.* **2009**, *1*, 482. (c) Hashmi, A. S. K. *Angew. Chem., Int. Ed.* **2010**, *49*, 5232. (d) Xiao, J.; Li, X. *Angew. Chem., Int. Ed.* **2011**, *50*, 7226.

(25) (a) Kusuma, H.; Yamabe, H.; Onizawa, Y.; Hoshino, T.; Iwasawa, N. *Angew. Chem., Int. Ed.* **2005**, *44*, 468. (b) Onizawa, Y.; Kusama, H.; Iwasawa, N. *J. Am. Chem. Soc.* **2008**, *130*, 802. (c) Hiroyuki, K.; Onizawa, Y.; Iwasawa, N. *J. Am. Chem. Soc.* **2006**, *128*, 16500. (d) Kusama, H.; Karibe, Y.; Onizawa, Y.; Iwasawa, N. *Angew. Chem., Int. Ed.* **2010**, *49*, 4269. (e) Kusama, H.; Karibe, Y.; Imai, R.; Onizawa, Y.; Yamabe, H.; Iwasawa, N. *Chem.—Eur. J.* **2011**, *17*, 4839.